Shallow Groundwater Monitoring at the American Electric Power Plant CO2 Product Validation Facility Monitoring and Verification: ENOS WP3 Workshop 26 April 2018

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## **Project Timeline**



## The Monitoring Program at the AEP PVF Site was a Comprehensive, Integrated Program



The PVF project at the AEP Mountaineer Power Plant consisted of a 20 MW CO<sub>2</sub> Capture and Storage System

CO<sub>2</sub> Capture and Injection took place from October 2009 - May 2011

Well network of 2 injection wells and 3 reservoir monitoring wells

Injection into two reservoirs, the Rose Run Sandstone and the Copper Ridge Dolomite





#### Site Layout Showing Well Proximities

Groundwater Monitoring Wells Surround the Deep Injection and Monitoring Wells.

Deep Monitoring Wells completed approx. 7,800 – 8,000 ft.

Injection Wells completed approx. 7,800 – 8,000 ft.

Direction of shallow groundwater flow





#### Monitoring parameters – shallow groundwater

Cations	Anions	Physical Parameters	Other <sup>(b)</sup>
Potassium Sodium Calcium Magnesium Iron Manganese Aluminum Barium Boron Lithium <sup>(b)</sup> Strontium Dissolved Silica	Chloride Sulfate Bromide Fluoride	pH <sup>(a)</sup> Alkalinity (Bicarbonate) Alkalinity (Carbonate) Total Dissplved Solids Specific gravity/ Density <sup>(b)</sup> Dissolved Organic Carbon Specific conductance <sup>(a)</sup> Temperature <sup>(a)</sup> Turbidity <sup>(a)</sup>	Stable hydrogen isotopes (D/H) Stable oxygen isotopes ( <sup>18</sup> O / <sup>16</sup> O) Stable carbon isotopes ( <sup>13</sup> C/ <sup>12</sup> C) Dissolved CO <sub>2</sub>

(a) Field parameter

(b) Optional parameter, may be done at AEP's discretion

#### How to detect a leak signal?

- Change in chemical parameters, e.g.
  - decreased pH caused by dissolution of CO<sub>2</sub>
  - increase in alkalinity dissolution of carbonate minerals by acidic fluids
  - increase in TDS mineral dissolution and increase in cations such as Ca<sup>+2</sup> and Mg<sup>+2</sup>
  - Increase in acid-soluble metals such as iron and manganese
- Isotopes
  - Stable oxygen and hydrogen isotopes (δ<sup>18</sup>O and δD) of shallow groundwater differ characteristically from deep brines.
  - Stable carbon isotopes (δ13C) of shallow groundwater (CO<sub>2</sub>) differ from δ13C of anthropogenic CO<sub>2</sub> from coal combustion



#### **Chemical Parameter Monitoring**

Several chemical parameters increased in concentration starting around the same time that injection started

Possible indicators of upward CO2 leakage?



#### Stable Oxygen and Hydrogen Isotopes

## Possible upward brine migration?



Stable oxygen and hydrogen isotopes were monitored in the groundwater and compared with deep brine pre-injection samples to detect upward migration of deep brine from injection reservoirs

Post baseline MW-16 (red squares) deviates from meteoric line where other shallow water samples plot.

#### Stable Carbon Isotope Monitoring



CO2 from coal combustion (-20 to -30‰)

No indicators of upward CO2 leakage?



Stable carbon

isotope activities

detected in all 4

wells fall between

- Further analysis indicated that the observed concentration increases in chemical indicator parameters were not due to upward migration of CO<sub>2</sub> from the injection reservoirs.
- Further analysis of the Stable oxygen and hydrogen isotopes indicated that the deviations in mw-16 are probably not due to upward migration of brine from the injection reservoirs.





#### Stable Oxygen and Hydrogen Isotopes

Possible upward brine migration?





MW-16 shows shift in ion concentrations after baseline period. But, post baseline

MW-16 not on mixing line with deep brine and baseline MW-16

Figure 9-10. Well MW-16 Piper Diagram of Pre-Injection and Post-Injection Shallow Groundwater Data (Equivalents) (Open symbols represent post-injection data. Deep brine fluid from well AEP-1 plotted for reference)



### Stable Oxygen and Hydrogen Isotope data

- The changes seen in post-injection Stable Oxygen and Hydrogen Isotope data from well MW-16 appears to be due to mixing of shallow aquifer water with a highly evaporated water, which has higher TDS and is isotopically significantly heavier than meteoric water. This could be due to a number of plant activities,
- Other possible sources of water and solutes around the plant were investigated to attempt to explain the geochemical changes observed in the shallow aquifer during post-injection sampling.



- Leaky pond hypothesis water from the nearby bottom ash ponds or wastewater ponds may have leached into the shallow groundwater and caused the observed composition changes at MW-16C
- 5 ponds were sampled and compared to groundwater data
- Mixing models show that major ion ratios are not consistent with a simple mixing ratio, and different proportions of pond water were required to simulate observed δ18O values than chloride concentrations.



## **Summary**

- It isn't easy to identify an impact to shallow groundwater, especially at industrial sites
- False positives are possible, especially with cation/anion data
- Use statistics to analyze data
- Isotope data can help explain apparent leakage signals
- If signals are detected, focus on disproving CO2/brine leakage rather than proving the cause, which may require additional data and much greater effort...and still not find the cause...



# Gaps in shallow groundwater monitoring methodology

 Need protocol(s) for analyzing sampling data to discern evidence for CO2 or brine leakage...

